GENERAL CHEMISTRY									
1	Course Title:	GENER/	AL CHEMISTRY						
2	Course Code:	KIM1079	OTR .						
3	Type of Course:	Compuls	sory						
4	Level of Course:	First Cyc	cle						
5	Year of Study:	1							
6	Semester:	1							
7	ECTS Credits Allocated:	3.00							
8	Theoretical (hour/week):	3.00							
9	Practice (hour/week):	0.00							
10	Laboratory (hour/week):	0							
11	Prerequisites:	None							
12	Language:	Turkish							
13	Mode of Delivery:	Face to t	face						
14	Course Coordinator:	Prof. Dr.	MEHMET HALUK TÜRKDEMİR						
15	Course Lecturers:								
16	Contact information of the Course Coordinator:	nturkel@uludag.edu.tr 0 224 294 17 30 Uludağ Üniversitesi Fen-Edebiyat Fakültesi Kimya Bölümü, Görükle/BURSA 16059							
17	Website:								
18	Objective of the Course:	The purpose of this course, learn the basic concepts of general chemistry in the content area and has learned to be able to use this information to provide in their field.							
19	Contribution of the Course to Professional Development:								
20	Learning Outcomes:								
		1	To learn the basic concepts and the methods in the field of general chemistry and apply this knowledge to the field in food engineering.						
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		8							
		9							
		10							
21	Course Content:								
107		Co	purse Content:						
	Theoretical		Practice						
1	Matter-Its properties and measurem	ent							
2	Atoms and atomic theory	otions							
3	Chemical compounds, Chemical rea	CTIONS							

4	Aqueous solution reactions																	
5	Gases																	
6	Thermochemistry																	
7	Midterm exam																	
8	The periodic table and some atomic properties																	
9	Che	Chemical Bonding I																
10	Acids and bases																	
11	Metals																	
12	Nonmetals																	
13	Organic chemistry																	
14	Chemical of the living state																	
22	Textbooks, References and/or Other Materials:							Ar Ha	1) General Chemistry I and II , Principles and Modern Applications. (Eighth edition) R. H. Petrucci , W.S. Harwood. 2) Principles of modern chemistry (Fourth edition), D. W. Oxtoby, H. P. Gillis, N. H. Nachtrieb, 1999.									
23	Ass	esme	ent															
TERM L	LEARNING ACTIVITIES NUMBE						W	WEIGHT										
Midterr	m Ex	am					1		40	0.00								
Activites							Number [Duration (hour)			Total Work Load (hour)				
Theore	etical						2)	10	100.00				3.00			42.00	
Practic										0				0.00			0.00	
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Homew	neworks								2				10.00			20.00		
Popierct	ects							10	100.00				0.00			0.00		
Field S	Studies								0				0.00			0.00		
Oliodtesre	amen exams								1			10.00	10.00			10.00		
Others	'S								0				0.00			0.00		
	al Exams								1			10.00	10.00			10.00		
	otal Work Load															92.00		
Total w	Γotal work load/ 30 hr														3.07			
ECTS (CTS Credit of the Course															3.00		
25	25 CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS																	
		PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16	
ÖK1		2	1	2	1	2	1	1	2	2	2	1	1	0	0	0	0	
			<u> </u>	 լ	earr	l nina C) bier	ctives		PQ: P	roars	ım Qıı	l alifica	tions	 S			
Conti ution Leve	rib 1 very low		earning Objective 2 low			Т		lium	4 High					y High				