	THERMODYNAMICS I										
1	Course Title:	THERM	ODYNAMICS I								
2	Course Code:	OTO2007									
3	Type of Course:	Compulsory									
4	Level of Course:	First Cyc	cle								
5	Year of Study:	2									
6	Semester:	3									
7	ECTS Credits Allocated:	5.00									
8	Theoretical (hour/week):	3.00									
9	Practice (hour/week):	0.00									
10	Laboratory (hour/week):	0									
11	Prerequisites:	None									
12	Language:	Turkish									
13	Mode of Delivery:	Face to face									
14	Course Coordinator:	Prof. Dr. ERHAN PULAT									
15	Course Lecturers:	Doç. Dr. NURETTİN YAMANKARADENİZ									
16	Contact information of the Course Coordinator:	pulat@uludag.edu.tr 0 224 2941982 Bursa Uludağ Üniversitesi, Mühendislik Fakültesi, Makina Müh. Bölümü, Oda No:217, Görükle, 16059, Bursa									
17	Website:										
18	Objective of the Course:	Teaching of fundamental concepts and laws of thermodynamics by engineering approach.									
19	Contribution of the Course to Professional Development:	Knowing performance limits of heat engines and having possible environmental effects of heat engines, refrigerators, and heat pumps. Constructing relations of thermodynamics with other engineering branches, social disciplines, and ecosystems.									
20	Learning Outcomes:										
	•	1	Explaining of thermodynamic concepts and laws by proper using of thermodynamic terminiology.								
		2	Doing distinction between pure substance and ideal gas, finding or determination of properties of substances.								
		3	Using of first and second law of thermodynamics in problem solving.								
		4	Doing proper assumptions to solve thermodynamic problems and obtaining thermodynamic data.								
		5	Knowing performance limits of heat engines. Having possible environmental effects of heat engines, refrigerators, and heat pumps.								
		6	Doing distinction between closed and open systems.								
		7	Constructing relations of thermodynamics with other engineering branches, social disciplines, and ecosystems.								
		8									
		9									
		10									
21	Course Content:										
		Co	ourse Content:								
Week Theoretical Practice											

1	Introduction to thermodynamics, define closed and open systems, properties processes and cycle.									
2	Pure substances, equilibrium diagrar thermodynamic tables, equations of s ideal gas equation of state.									
3	Work and Heat. Moving boundary wo	ork.								
4	Nature of work and heat, heat transfe and power.	er rate								
5	First law of thermodynamics, internal enthalpy, and specific heats.	energy,								
6	First law analysis, internal energy, er and specific heats of ideal gases.	nthalpy								
7	Second law of thermodynamics. Hea sources. Carnot's principle. Heat eng Thermal efficiency. Heat pumps and refrigerators. Coefficient of performan Clausius's and Kelvinb Planck's state	jines. nce,								
8	Mid-term exam.	2								
9	Reversible and irreversible processes cycle. Thermodynamic temperature s	scale.	L							
10	Entropy. Clausius's inequalty. Proper diagrams related to entropi. Increase entropi principle.									
11	Entropy change of pure substance. E change of ideal gases. Reversible an									
Activites				Number	Duration (hour)	Total Work Load (hour)				
Theore	ticalumes. Flow work. Second law of			14	3.00	42.00				
Practic	als/Labs			0	0.00	0.00				
Self stu	Canulation of the for SSSF process	ses.		14	2.00	28.00				
Homew	vorks			0	0.00	0.00				
Project 14	Uniform State Uniform Flow (USUF)			0	0.00	0.00				
Field S	tudies			0	0.00	0.00				
	Lexams Textbooks. References and/or Other		1	, Mühendislik Termodir	40.00 amidin Temelleri. (40.00 Jilt 1. R.				
Others				0	0.00	0.00				
Final E			2	nkara. Mühendislik Yaklasım	40.00 ivia Termodinamik.					
Total Work Load				ayıncılık ə. basını, eki		150.00				
Total w	ork load/ 30 hr			Çözümlü Problemlerle	Termodinamik, A.	500 Ozturk, A.				
ECTS Credit of the Course						5.00				
		 Atilgan, Pamuk Ofset, 1985, İstanbul. 5. Elements of Environmental Engineering: Thermodynamics and Kinetics, 2nd Ed., K. Valsaraj, Lewis Publishers, Inc., 2000. 6. Thermodynamics of Atmospheres and Oceans, J. A. Curry and P. J. Webster, Academic Press, 1999. 								
23		NUMBE	1.4							
		WEIGHT								
Midterm Exam 1				40.00						
Quiz 0				0.00						
	work-project	0	0.00							
Final Exam 1				60.00						

								4.0								
Total 2						_	100.00									
Contribution of Term (Year) Learning Activities to Success Grade							40.	40.00								
Contribution of Final Exam to Success Grade						60.	60.00									
Total						10	100.00									
Measurement and Evaluation Techniques Used in the Course					ne Cla	Classical Written Midterm and Final Exam										
24	ECTS / WORK LOAD TABLE															
25	25 CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS															
	PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16
ÖK1	5	5	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ÖK2	5	5	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ÖK3	5	5	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ÖK4	5	5	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ÖK5	5	5	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ÖK6	5	5	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ÖK7	5	5	0	0	0	0	0	0	0	0	0	0	0	0	0	0
LO: Learning Objectives PQ: Program Qualifications																
Contrib 1 very low ution Level:				2 low		3 Medium			4 High			5 Very High				