	В	ЮСН	EMISTRY I					
1	Course Title:	BIOCHEMISTRY I						
2	Course Code:	VET1008						
3	Type of Course:	Compulsory						
4	Level of Course:	First Cycle						
5	Year of Study:	1						
6	Semester:	2						
7	ECTS Credits Allocated:	4.00						
8	Theoretical (hour/week):	3.00						
9	Practice (hour/week):	2.00						
10	Laboratory (hour/week):	0						
11	Prerequisites:	-						
12	Language:	Turkish						
13	Mode of Delivery:	Face to face						
14	Course Coordinator:	Prof. Dr. MELTEM TANRIVERDİ						
15	Course Lecturers:	Prof.Dr. Ümit Polat, Prof.Dr. Nazmiye Güneş						
16	Contact information of the Course Coordinator:	melcetin@uludag.edu.tr 0 224 2941181 U.Ü.Veteriner Fakültesi Biyokimya ABD						
17	Website:							
18	Objective of the Course:	Comprehension of biological molecules, carbohydrates, lipids, proteins, enzymes, minerals, and their classification and basic biochemical knowledge about their physical, chemical and functional properties						
19	Contribution of the Course to Professional Development:	Being able to define carbohydrates, lipids, and proteins which constitute the energy sources and building blocks of the living organisms,Comprehension of basic knowledge that may be essential to other courses in veterinary medicine,Being able to disseminate knowledge gained about basic biochemical subjects both verbally and in writing.						
20	Learning Outcomes:							
	•	1	Comprehension of biophysical processes that are essential to survival of the living organisms.					
		2	Being able to define carbohydrates, lipids, and proteins which constitute the energy sources and building blocks of the living organisms					
		3 Comprehension of classification of enzymes and mir their properties and functions; being able to assess situations characterized by their excess and deficien						
		4	Being able to comprehend laboratory safety regulations and being able to gain qualification to work in the laboratory.					
		5	Being able to define basic reactions utilizing biochemical substances and apply them in the laboratory.					
		6 Being able to interpret results obtained from studies in th laboratory.						
		 Comprehension of basic knowledge that may be essential to other courses in veterinary medicine. 						
		8	Being able to disseminate knowledge gained about basic biochemical subjects both verbally and in writing.					
		9						

		10								
21	Course Content:									
		Co	urse Content:							
Week	Theoretical		Practice							
1	Description and subjects of biochemis functions, distribution and metabolisn water, properties of biochemical solut	n of cell	Introduction to biochemistry laboratory, the working principles in the laboratory, laboratory accidents, first aid informations							
2	Diffusion, osmotic pressure, dialysis, tension, absorption, freezing point depression, hydrogen ion concentrati isotopes, the definition and classificat carbohydrates, aldoses and ketoses, formulation and D and L forms of monosaccharides and epimerism.	on and tion of	Kinds of biochemical analysis, learning how to use the equipments and materials used for the qualitative analyses							
3	Asymmetric carbon atom, stereoisom optical activity, enantiomers, structure hemiacetal, 3 and a isomers, mutarot specific rotation	e of the	Molisch, Seliwanoff and Moore experiments, their principles and interpretations							
4	Osazone formation, the effects of acialkalines on monosaccharides, enolis reduction and oxidation products of s amino and deoxy sugars, phosphate derivative	sation, ugars,	Fehling, Benedict and Nylander experiments, their principles and interpretations							
5	Glycosides, properties of monosacch and disaccharides, maltose and treha type glycosidic binding, properties of lactose, saccharose and cellobiose	alose maltose,								
Activit			Numbe	r	Duration (hour)	Total Work Load (hour)				
	Description, classification and proper lipids, saturated, unsaturated and ess als/Labs	ties of sential	Hydrolysis structure of 14	of saccharos saccharose	в hyestigating the reaction 2.00	properties and 28.00				
Self stu	properties, simple and complex give dy and preperation	enas	10		3.00	30.00				
Homew			0		0.00	0.00				
Project	neutral lipids, the simple and conjuga	ited	0		0.00	0.00				
Field S			0		0.00	0.00				
Midterr	aexawaxes,terpenes, carotenoids, bil	e acids.	in l unsatura	ted fatty acid	5.09terification, Sa	Koo0/0ski				
Others			1		5.00	5.00				
Final E	tems The physical properties, chemical rea	actions	Xanthoprot	ein, Biuret ar	d Ninhydrine expe	iments, their				
	Vork Load					120.00				
Total w	ork load/ 30 hr The formation of pentid bound, the st	ructure	Pettenkofe	· Hav Gmeli	h and Rosin experi	4.00 ments their				
ECTS	Credit of the Course					4.00				
	denaturation of protein molecules, sir conjugated proteins	nple and								
11	Nucleic acids, structures, properties a types of DNA and RNA, chromoprote heme and phorphyrins, ETS's protein bilirubin and biliverdin	ins,	Precipitation of proteins with ammonium sulphate and heat							
12	Properties of minerals, macromineral microelements, presence in the blood deficiencies of minerals and malfunct	d, ions	Precipitation of proteins with mineral acids, heavy metals and alkaloids							
13	Enzyme, catalyser, apoenzyme, prosthetic group, the structure and properties of enzymes, the mechanisms of enzyme reactions, action way, nomenclature of enzymes									

14	affe	assification of enzymes, the factors ecting enzyme activity, enzyme inhibition, osteric enzymes, coenzymes								The determination of catalase in the blood and liver								
22		xtbooks, References and/or Other aterials:																
23	Asse	esme	ent															
TERMI	I LEARNING ACTIVITIES							WE	WEIGHT									
Midterm Exam 1							30	.00										
Quiz	luiz						1		10	.00								
Home	work-	proje	ect				0)	0.0	0.00								
Final E	xam						1		60	60.00								
Total	al 3							10	100.00									
Contribution of Term (Year) Learning Activities to Success Grade							40	40.00										
Contrib	Contribution of Final Exam to Success Grade						60	60.00										
Total									10	0.00								
Measu Course 24	e					hnique		d in th	ie Qu	iizz, M	id exa	m, Fina	l exam					
		107																
25)			CON	IRIE	SUTIC				_			510	PROC	GRAM	WE		
		PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16	
ÖK1	4	5	1	4	2	1	3	1	1	1	1	1	1	0	0	0	0	
ÖK2		4	1	4	2	1	4	1	1	1	1	1	1	0	0	0	0	
ÖK3		5	1	4	5	2	2	1	1	1	1	1	1	0	0	0	0	
ÖK4		4	1	5	3	3	2	1	1	1	1	1	1	0	0	0	0	
ÖK5		4	1	4	4	2	2	1	1	1	1	1	1	0	0	0	0	
ÖK6	•	4	1	4	4	2	1	1	1	1	1	1	1	0	0	0	0	
ÖK7		4	1	4	1	1	5	1	1	1	1	1	1	0	0	0	0	
ÖK8		4	1	4	1	4	1	1	1	1	4	1	1	0	0	0	0	
			l	LO: L	earr	ning C	Dbjec	tives	s F	Q: P	rogra	am Qu	alifica	tions	5		-	
utio	Contrib 1 very low ution Level:		2 low 3 M			Medi	edium 4 High			5 Very High								