SAN	IPLE PREPARATION	METH	ODS IN ANALYTICAL CHEMISTRY				
1	Course Title:	SAMPLE PREPARATION METHODS IN ANALYTICAL CHEMISTRY					
2	Course Code:	KIM5007					
3	Type of Course:	Optional					
4	Level of Course:	Second Cycle					
5	Year of Study:	1					
6	Semester:	1					
7	ECTS Credits Allocated:	6.00					
8	Theoretical (hour/week):	3.00					
9	Practice (hour/week):	0.00					
10	Laboratory (hour/week):	0					
11	Prerequisites:	None					
12	Language:	Turkish					
13	Mode of Delivery:	Face to face					
14	Course Coordinator:	Prof. Dr. BELGİN İZGİ					
15	Course Lecturers:	Prof. Dr. Belgin İZGİ Prof. Dr. M. H. TÜRKDEMİR Prof. Dr. Saliha ŞAHİN Prof. Dr. Elif TÜMAY ÖZER Doç. Dr. Ümran SEVEN ERDEMİR Doç. Dr. Önder AYBASTIER					
16	Contact information of the Course Coordinator:	belgin@uludag.edu.tr 0 224 29 41 728					
17	Website:						
18	Objective of the Course:	The aim of the course is to make the students gain the basic knowledge and skills about different sample preparation methods for different matrices. So that students may encounter in graduate studies at analytical analysis methods is expected to be successful.					
19	Contribution of the Course to Professional Development:	Helps basic chemistry knowledge develop new approaches for real sample					
20	Learning Outcomes:						
		1	Optimize the use of methods of sample preparation on analytical thinking.				
		2	Enhance the ability to approach on different matrices, depending on the method of analysis for selection of the desired analyte.				
		3	Gain the skill assessment for evaluation of the uncertainties of the process until the outcome of the analysis.				
		4	Evaluate the performance criteria and develops the capability of responding according to the method of analysis.				
		5	Learn what parameters to use for transport of sample analysis results, gains perspective.of the GLP (Good Laboratory Practices)				
		6					
		7					
		8					
		9					
		10					

21	Course Content:										
	Course Content:										
Week	Theoretical		Practice								
1	General introduction, methods of san preparation and analytical examination relationship between performance an	nple on of the id time									
2	Examination of the sample preparation techniques (homogenization, dissolution methods (wet-dry), extraction methods liquid (LLE), solid phase (SPE)), concentration, analysis, investigation	on tin ds (liquid- steps									
3	Relating to the subject with emphasis performance characteristics	son									
4	The most basic step in terms of samp preparation conditions and the impor sampling, the sampling design	ole tance of									
5	As a storage container for the sample conditions, etc. The relationship betw products and the matrix	e storage veen									
6	Volatile and nonvolatile storage of sp of species in regard to evaluation of approaches and pollutants	ecimens									
7	According to differences in the matrix	in terms									
Activit	es		Number	Duration (hour)	Total Work Load (hour)						
Th g ore	Madterm Exam + repetition of previou	s issues	14	3.00	42.00						
Practica	als/Labs		0	0.00	0.00						
Self stu	Imixtures assessment and green cher dy and preperation (green chemistry) relationship	nistry	14	4.00	56.00						
Homew	vorks		0	0.00	0.00						
Project	principle of LLE and SPE methods		0	0.00	0.00						
Field St	tudies		0	0.00	0.00						
Midtern 13	SCE (Super critical extraction) and th	e use of	1	30.00	30.00						
Others			0	0.00	0.00						
Final E	assessments relating to applications	na tha	1	52.00	52.00						
Total W	/ork Load				180.00						
Total w	one for the infesearch about desired	analyte			6.00						
ECTS (Credit of the Course				6.00						
	Materials:		 [2] www.instrumentalchemistry.com/sampleprep [3] Sample Preparation Techniques in Analytical Chemistry, J. D. WINEFORDNER [4] Sample Preparation for Trace Element Analysis, Z. Mester, J. Sturgeon, 2003, Elsevier 								
23	Assesment										
TERM L	EARNING ACTIVITIES	NUMBE R	WEIGHT								
Midtern	n Exam	1	20.00								
Quiz		0	0.00								
Home v	work-project	1	20.00								
Final E	xam	1	60.00								
Total		3	100.00								
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Contribution of Term (Year) Learning Activities to Success Grade	40.00
Contribution of Final Exam to Success Grade	60.00
Total	100.00
Measurement and Evaluation Techniques Used in the Course	Relative evaluation is applied.

24 ECTS / WORK LOAD TABLE

25	CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS															
	PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16
ÖK1	5	0	0	0	5	0	0	0	0	5	0	0	0	0	0	0
ÖK2	5	0	4	0	5	0	0	0	0	0	0	0	0	0	0	0
ÖK3	5	0	4	0	5	0	0	0	0	0	0	0	0	0	0	0
ÖK4	5	0	4	0	5	0	0	0	0	0	0	0	0	0	0	0
ÖK5	5	0	4	0	5	0	0	0	0	0	0	0	0	0	0	0
LO: Learning Objectives PQ: Program Qualifications																
Contrib ution Level:	ntrib 1 very low 2 lov ion vel:			2 Iow		3 Medium			4 High			5 Very High				