	GENERAL CHEMISTRY									
1	Course Title:	GENERA	GENERAL CHEMISTRY							
2	Course Code:	KIM1079								
3	Type of Course:	Compuls	sory							
4	Level of Course:	First Cyc	le							
5	Year of Study:	1								
6	Semester:	1								
7	ECTS Credits Allocated:	3.00								
8	Theoretical (hour/week):	2.00								
9	Practice (hour/week):	0.00								
10	Laboratory (hour/week):	2								
11	Prerequisites:	None								
12	Language:	English								
13	Mode of Delivery:	Face to f	ace							
14	Course Coordinator:	Prof. Dr.	ASIM OLGUN							
15	Course Lecturers:									
16	Contact information of the Course Coordinator:	mesutgo 0 224 27	udağ Üniversitesi Fen-Edebiyat Fakültesi Kimya Bölümü							
17	Website:									
18	Objective of the Course:	The aim of this course is to explain basic concepts of chemistry and help to gain basic knowledge that students will need to other courses of chemistry.								
19	Contribution of the Course to Professional Development:	Learns chemical substances and their behavior. Understands chemical calculations.								
20	Learning Outcomes:									
		1	Learns the basic concepts of chemistry.							
		2	Comments the the basic concepts of chemistry.							
		3	Analyses the basic concepts of chemistry.							
		4	Performs the basic concepts of chemistry in the laboratory.							
		5								
		6								
		7								
		9								
		10								
21	Course Content:	10								
21	Course Content.	Co	ourse Content:							
Week	Theoretical		Practice							
1	Chemical kinetics									
2	Chemical equilibrium									
3	Acids and bases									
4	Solubility and complex ion equilibriur	n								
5	Solubility and complex ion equilibriur									

7	Flectroc	hamie	trv/														
_	Electrochemistry Popositing courses																
	Repeating courses																
	Main group elements																
	Transition elements																
	Complex ions and coordination compounds																
	Nuclear chemistry																
	Organic																
14	Chemistry of the living state																
							General Chemistry , Principles and Modern Applications. Petrucci Harwood.										
23	Assesm	ent															
TERM L	EARNING	ACTI	VITIES	;			UMBE	WE	EIGHT								
Midterm	Evam					1 1		40	40.00								
Quiz	LAAIII					0		0.0									
	ork-proj	act				0		0.0									
		U UI						_									
	al Exam 1							60.00									
	Total 2						100.00										
Contribution of Term (Year) Learning Activities to Success Grade																	
Activites							Number Duration (hour)					Total Work Load (hour)					
Megret	ical ent ar	nd Eva	aluatio	n Tec	hniaue	s Use	d in the	Hc	14 mewo	rks and	d writter	exam:	 S		28.00		
Measurement and Evaluation Techniques Used in the Practicals/Labs							`					28.00					
Sensulty and preperation LOAD TABLE							14			1.00			14.00				
Homeworks						(6 1.00			6.00							
Projects	Projects						-	0			0.00			0.00			
Field St	Field Studies							0			0.00			0.00			
Midterm	lidterm exams							ŀ	1			10.00			10.00		
Others								(0			0.00			0.00		
Final Ex								1	1			10.00			10.00		
Total W	ork Load	d													96.00		
Total wo	I work load/ 30 hr														3.20		
ECTS C	S Credit of the Course													3.00			
25 CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS																	
	PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1	PQ11	PQ12	PQ1	PQ14	PQ15	PQ16	
ÖK1	3	2	2	3	4	4	3 4	1	1	4	0	0	0	0	0	0	
ÖK2	3	2	2	3	4	4	3 4	1	1	4	0	0	0	0	0	0	
ÖK3	4	3	3	4	5	5	4 5	5	1	5	0	0	0	0	0	0	
ÖK4	5	4	5	4	4	5	4 5	5	1	5	0	0	0	0	0	0	
LO: Learning Objectives PQ: Program Qualifications																	

Entropy and free energy

Contrib	1 very low	2 low	3 Medium	4 High	5 Very High
ution					
Level:					