

# INORGANIC CHEMISTRY I

1	Course Title:	INORGANIC CHEMISTRY I	
2	Course Code:	KIM2007	
3	Type of Course:	Compulsory	
4	Level of Course:	First Cycle	
5	Year of Study:	2	
6	Semester:	3	
7	ECTS Credits Allocated:	5.00	
8	Theoretical (hour/week):	4.00	
9	Practice (hour/week):	0.00	
10	Laboratory (hour/week):	0	
11	Prerequisites:	There is no course prerequisite.	
12	Language:	Turkish	
13	Mode of Delivery:	Face to face	
14	Course Coordinator:	Prof. Dr. Veyssel Turan Yılmaz	
15	Course Lecturers:	Yok	
16	Contact information of the Course Coordinator:	vtyilmaz@uludag.edu.tr 02242941749	
17	Website:		
18	Objective of the Course:	To teach the basic concepts and use these concepts in Inorganic chemistry.	
19	Contribution of the Course to Professional Development:	Make use of theoretical and practical knowledge acquired in the field of Inorganic chemistry.	
20	Learning Outcomes:		
		1	Knows the structure of the atom
		2	Use the periodic chart,
		3	Examine the general properties of elements
		4	Examine the types of chemical bonding
		5	Determine the symmetry elements and groups of buildings,
		6	Examine theories explaining the structure and shape of the molecules,
		7	
		8	
		9	
		10	
21	Course Content:		
		<b>Course Content:</b>	
Week	Theoretical	Practice	
1	Atomic structure		
2	Many-electron structures, Periodic properties of elements		
3	Atom and ion size, ionization energy, electron affinity, and polarisation		

4	Electronegativity, Dipole moment, metallic character				
5	Bond types, the oxidation step, the formal charge				
6	Lewis formulas, resonance, metallic bond				
7	Symmetry elements and symmetry operations, point groups				
8	Symmetry classification of transactions, character tables				
9	Repetition of previous lessons and midterm				
10	Symmetry and polarization, symmetry and chirality				
11	Lewis theory, the Octet Rule				
12	The thrust of the valence shell electron pair rule				
13	Valence-bond theory, Hybridity				
14	Molecular orbital theory				
22	Textbooks, References and/or Other Materials:				
23	Assesment				
TERM LEARNING ACTIVITIES		NUMBER	WEIGHT		
Midterm Exam		1	40.00		
Activites			Number	Duration (hour)	Total Work Load (hour)
Final Exam		1	60.00	4.00	56.00
Practicals/Labs			0	0.00	0.00
Contribution of Preparation Learning Activities to Success Grade		40.00	4.00	4.00	56.00
Homeworks			0	0.00	0.00
Contribution of Final Exam to Success Grade		60.00	0.00	0.00	0.00
Field Studies			0	0.00	0.00
Measurements and Evaluation Techniques Used in the Written exams and multiple choice tests			20.00	20.00	20.00
Others			0	0.00	0.00
24. ECTS/ WORK LOAD TABLE					
Final Exams			1	20.00	20.00
Total Work Load					172.00
Total work load/ 30 hr					5.07
ECTS Credit of the Course					5.00

25	CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS															
	PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ10	PQ11	PQ12	PQ13	PQ14	PQ15	PQ16
ÖK1	5	5	5	3	4	3	4	1	1	2	3	4	2	0	0	0
ÖK2	5	5	5	3	4	3	4	1	1	2	3	4	2	0	0	0
ÖK3	5	5	0	3	4	3	4	1	1	2	3	4	2	0	0	0
ÖK4	3	3	3	3	3	3	4	1	1	2	2	2	2	0	0	0

ÖK5	3	3	3	3	3	3	4	1	1	2	2	2	2	0	0	0
ÖK6	3	3	3	3	3	3	4	1	1	2	2	2	2	0	0	0
LO: Learning Objectives    PQ: Program Qualifications																
Contribution Level:	1 very low		2 low		3 Medium		4 High		5 Very High							