	ADVANC	ED TH	IERMODYNAMICS					
1	Course Title:	ADVANCED THERMODYNAMICS						
2	Course Code:	MAK6213						
3	Type of Course:	Optional						
4	Level of Course:	Third Cycle						
5	Year of Study:	1						
6	Semester:	1						
7	ECTS Credits Allocated:	6.00						
8	Theoretical (hour/week):	3.00						
9	Practice (hour/week):	0.00						
10	Laboratory (hour/week):	0						
11	Prerequisites:	None						
12	Language:	Turkish						
13	Mode of Delivery:	Face to face						
14	Course Coordinator:	Prof. Dr. RECEP YAMANKARADENİZ						
15	Course Lecturers:	Prof. Dr. Recep Yamankaradeniz						
16	Contact information of the Course Coordinator:	pulat@uludag.edu.tr , 0 224 2941982 Uludağ Üniversitesi, Makina Mühendisliği Bölümü, Oda No: 217, Görükle, 16059, Bursa.						
17	Website:							
18	Objective of the Course:	The objectives of this course are - to reinforce the students grasp of classical thermodynamics, - to present topics in classical thermodynamics that are often not adequately covered in undergraduate courses such as availability analysis and 2nd law efficiency.						
19	Contribution of the Course to Professional Development:							
20	Learning Outcomes:							
		1	Understanding of the basic thermodynamic definitions and concepts.					
		2	Evaluation of the thermodynamic state and properties for pure substances and ideal gases.					
		3	Evaluation of the work and heat transfer of processes.					
			Application of the principles of conservation of mass and the 1st Law of Thermodynamics to closed and open systems.					
			Application of the 2nd Law of Thermodynamics to determine ultimate cycle performance and efficiency limitations. The availability (exergy) analysis.					
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		10						
21	Course Content:							
		Co	ourse Content:					
Week	Theoretical Practice							

1 Review of basic thermodynamic definitions and concepts. 2 Temperature, thermal equilibrium and zeroth law of thermodynamics. 3 Thermodynamic state and properties of pure substance and ideal gases. Principles of corresponding states. 4 Properties of ideal gases. Equations of state such as Van der Waals, Redlich-Kwong, and Benedect-Webb-Rubin. 5 Work and heat. Moving boundary work and other work types. First law of								
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other work types. First law of								
thermodynamics, internal energy and enthalpy.								
6 Open systems, first law and work types in open systems.								
7 Application of first law to ideal gases. Variable specific heats of ideal gases.								
8 Course Review and Midterm exam								
9 Second law and Carnot cycle. Heat engine, refrigeration machine and heat pumps.								
10 Entropy. Internal and external irreversibilities. TdS relations.								
11 Reversible work, actual work, usefull work and lost work.								
12 Availability (Exergy) analysis. Second law								
Theoretical 14 3.00	Load (hour) 42.00							
Practicals/Labs 0.00	Engineoro 0.00							
Self study and preperation 2 A Bejan. Advanced Eggingeering Thermore								
Lipha Wilow and Sana Inf. 1000	72.00							
Projects Engineers. Arnold, 1997 _{0.00}	0.00							
I/I D. Vamankaradoniz, Fundamontals of En	dipooring 0.00							
Midtern exams 5 같. A. Çengel and M. Botego Thermodyna	niesioAn							
Others 0 0.00	0.00							
Final Exams 1 22.00	22.00							
Total Work Load	196.00							
Terrivi Learning ACTIVITIES INDIVIBE WEIGHT	6.53							
	6.00							
Quiz 0 0.00	0.00							
	25.00							
Home work-project 6 25.00	50.00							
Final Exam 1 50.00								
Final Exam150.00Total8100.00Contribution of Term (Year) Learning Activities to50.00								
Final Exam150.00Total8100.00Contribution of Term (Year) Learning Activities to Success Grade50.00								
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25	CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS															
	PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16
ÖK1	4	4	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ÖK2	5	5	4	3	0	0	0	0	0	0	0	0	0	0	0	0
ÖK3	4	4	3	0	0	0	0	0	0	0	0	0	0	0	0	0
ÖK4	4	4	3	0	0	0	0	0	0	0	0	0	0	3	0	0
ÖK5	4	5	4	0	0	0	0	0	0	0	0	0	0	3	0	0
		l	LO: L	earr	ning C	Dbjed	tive	s P	Q: P	rogra	ım Qu	alifica	tions	5		
Contrib ution Level:	ution			2 low			3 Medium			4 High			5 Very High			