GENERAL CHEMISTRY									
1	Course Title:	GENER/	AL CHEMISTRY						
2	Course Code:	KIM1079	OTR .						
3	Type of Course:	Compuls	sory						
4	Level of Course:	First Cyc	cle						
5	Year of Study:	1							
6	Semester:	1							
7	ECTS Credits Allocated:	3.00							
8	Theoretical (hour/week):	3.00							
9	Practice (hour/week):	0.00							
10	Laboratory (hour/week):	0							
11	Prerequisites:	None							
12	Language:	Turkish							
13	Mode of Delivery:	Face to t	face						
14	Course Coordinator:	Prof. Dr.	MEHMET HALUK TÜRKDEMİR						
15	Course Lecturers:								
16	Contact information of the Course Coordinator:	nturkel@uludag.edu.tr 0 224 294 17 30 Uludağ Üniversitesi Fen-Edebiyat Fakültesi Kimya Bölümü, Görükle/BURSA 16059							
17	Website:								
18	Objective of the Course:	The purpose of this course, learn the basic concepts of general chemistry in the content area and has learned to be able to use this information to provide in their field.							
19	Contribution of the Course to Professional Development:								
20	Learning Outcomes:								
		1	To learn the basic concepts and the methods in the field of general chemistry and apply this knowledge to the field in food engineering.						
		2							
		3							
		4							
		5							
		6							
		7							
		8							
		9							
		10							
21	Course Content:								
107	<del></del>	Co	purse Content:						
	Theoretical		Practice						
1	Matter-Its properties and measurem	ent							
2	Atoms and atomic theory	otions							
3	Chemical compounds, Chemical rea	CTIONS							

4	Aqueous solution reactions							
5	Gases							
6	Thermochemistry							
7	Midterm exam							
8	The periodic table and some atomic							
0	properties							
9	Chemical Bonding I							
10	Acids and bases							
11	Metals							
12	Nonmetals							
13	Organic chemistry							
14	Chemical of the living state							
22	Textbooks, References and/or Other		General Chemistry I and II , Principles and Modern					
22	Materials:		Applications. (Eighth edition) R. H. Petrucci , W.S. Harwood.					
			2) Principles of modern chemistry (Fourth edition), D. W. Oxtoby, H. P. Gillis, N. H. Nachtrieb, 1999.					
23	Assesment							
TERM L	EARNING ACTIVITIES	NUMBE R	WEIGHT					
Midterr	n Exam	1	40.00					
Quiz		0	0.00					
Home v	work-project	0	0.00					
Final E	Final Exam 1		60.00					
Total		2	100.00					
Contribution of Term (Year) Learning Activities to Success Grade			40.00					
Contrib	ution of Final Exam to Success Grade	9	60.00					
Total			100.00					
Measu Course	rement and Evaluation Techniques Us	sed in the						
24	ECTS / WORK LOAD TABLE							

Activites	Number	Duration (hour)	Total Work Load (hour)
Theoretical	14	3.00	42.00
Practicals/Labs	0	0.00	0.00
Self study and preperation	2	5.00	10.00
Homeworks	2	10.00	20.00
Projects	0	0.00	0.00
Field Studies	0	0.00	0.00
Midterm exams	1	10.00	10.00
Others	0	0.00	0.00
Final Exams	1	10.00	10.00
Total Work Load			92.00
Total work load/ 30 hr			3.07
ECTS Credit of the Course			3.00

25	CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS															
	PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16
ÖK1	2	1	2	1	2	1	1	2	2	2	1	1	0	0	0	0
LO: Learning Objectives PQ: Program Qualifications																
Contrib 1 very low ution Level:			2 low		3 Medium			4 High			5 Very High					