THERMODYNAMICS I										
1	Course Title:	THERM	HERMODYNAMICS I							
2	Course Code:	MAK200	17							
3	Type of Course:	Compuls	sory							
4	Level of Course:	First Cyc	cle							
5	Year of Study:	2								
6	Semester:	3								
7	ECTS Credits Allocated:	5.00								
8	Theoretical (hour/week):	3.00								
9	Practice (hour/week):	0.00								
10	Laboratory (hour/week):	0								
11	Prerequisites:	None								
12	Language:	Turkish								
13	Mode of Delivery:	Face to	face							
14	Course Coordinator:	Prof. Dr.	RECEP YAMANKARADENİZ							
15	Course Lecturers:	Prof. Dr. Atakan Avcı Yrd. Doç. Dr. Erhan Pulat								
16	Contact information of the Course Coordinator:	recep@uludag.edu.tr, 0224 2941969 Uludağ Üniversitesi Mühendislik – Mimarlık Fakültesi Makine Mühendisliği Bölümü 16059 Görükle/BURSA								
17	Website:									
18	Objective of the Course:	This course is aimed to teach the basic laws of thermodynamics and to apply these laws to thermodynamics systems.								
19	Contribution of the Course to Professional Development:									
20	Learning Outcomes:									
		1	Comprehension of the thermodynamic concepts and laws by using the thermodynamic terminology properly.							
		2	Distinction between pure substance and ideal gas, and evaluation of the properties of the substances.							
		3	Using the first and second law of thermodynamics to solve problems.							
		4	Constitution of appropriate assumptions and obtaining thermodynamic data necessary to solve thermodynamic problems.							
		5	Determination of the limits of the performance of the thermal engines.							
		6	Distinction between closed-systems and steady-flow processes.							
		7	Comprehension of the relationship of thermodynamics to other engineering and non-engineering disciplines.							
		8								
		9								
		10								
21	Course Content:									
		Co	ourse Content:							
Week	Veek Theoretical Practice									

1	Introduction to thermodynamics, defining closed and open systems, properties	nitions, ,							
2	Pure substance, equilibrium diagram thermodynamic tables, equations of s ideal gas equation of state.	s, state,							
3	Work and heat. Moving boundary wo	rk.							
4	Constitution of work and heat. Heat t rate and power.	ransfer							
5	First law of thermodynamics, internal enthalpy and specific heats.	energy,							
6	Analysis of first law of thermodynami internal energy, enthalpy and specific ideal gas.	cs, c heats of							
7	The second law of thermodynamics. energy reservoirs. Heat engines. The efficiency. Kelvin-Planck Statement.	Thermal ermal							
8	Repeating courses and midterm examined and the second s	n							
9	Continuation of second law of thermodynamics. Refrigerators and h pumps. Coefficient of performance. C Statement. Reversible and irreversibl processes. The Carnot cycle. Thermodynamic temperature scale.	neat Clausius Ie							
10	Entropy. The Clausius inequality. Pro diagrams involving entropy. Its relation Principle of the increase of entropy	operty ons.							
Activit	es			Number	Duration (hour)	Total Work Load (hour)			
Theore	Irreversible processes of pure substa	nces	Π	14	3.00	42.00			
Practic	als/Labs			0	0.00	0.00			
Self stu	first have the same diameter of the same diameter o	systems.		14	3.00	42.00			
Homew	vorks			0	0.00	0.00			
Pr <b>¢j3</b> ect	Steady-state and steady flow proces	ses.		0	0.00	0.00			
Field S	tudies			0	0.00	0.00			
Midtern	devares.	3		1	22.00	22.00			
Others				2	8.00	16.00			
Final E	Kams Textbooks, References and/or Other		1	1 Mühendislik Termodir	28.00 Jamičin Temelleri (	28.00 Lilt 1 R			
Total W	/ork Load					150.00			
Total w	ork load/ 30 hr		Ai 2	nkara. <u>Mühendislik Yaklasım</u>	uvla Termodinamik	5,00 Y A Cengel			
ECTS (	Credit of the Course					5.00			
			<ul> <li>3- Çözümlü Problemlerle Termodinamik, A. Öztürk, A. Kılıç, 3. Basım, Çağlayan Kitapevi, 1993, İstanbul.</li> <li>4- Çözümlü Termodinamik Problemleri, A.N. Eğrican, H. Atılgan, Pamuk Ofset, 1985, İstanbul.</li> <li>5- Termodinamik Cilt 1, Termodinamiğin Temel Yasaları, A.R. Büyüktür, U.Ü. Basımevi, 1982, Bursa.</li> <li>6- Fundamentals of Classical Thermodynamics, G.J. Won Wylen, R.E. Sonntag, 3rd ed. SI Version, John Wiley and Sons, 1985, Singapore.</li> <li>7- Fundamentals of Engineering Thermodynamics, M.J. Moran, H.N. Shapiro, 3th Ed. SI Version, John Wiley and Sons, 2000, Singapore.</li> </ul>						
23	Assesment								
TERML	EARNING ACTIVITIES	NUMBE R	w	EIGHT					

Midterm Exam 1						30.	30.00										
Quiz 2								20.	20.00								
Home work-project 0							0.0	0.00									
Final Exam 1							50.	50.00									
Total 4							10	100.00									
Contribution of Term (Year) Learning Activities to Success Grade							50.	50.00									
Contribution of Final Exam to Success Grade							50.	50.00									
Total							10	100.00									
Measurement and Evaluation Techniques Used in the Course							ne										
24 ECTS / WORK LOAD TABLE																	
25 CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS																	
	PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16	
ÖK1	0	4	0	0	0	0	0	3	0	0	0	0	0	0	0	0	
ÖK2	4	4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
ÖK3	4	5	3	0	0	0	0	0	0	0	0	0	0	0	0	0	
ÖK4	4	5	3	0	0	0	0	0	0	0	0	0	0	0	0	0	
ÖK5	2	4	4	0	0	0	0	0	0	0	0	0	0	2	0	0	
ÖK6	2	5	4	0	0	0	0	0	0	0	0	0	0	0	0	0	
ÖK7	3	2	3	0	0	0	0	0	0	0	0	0	0	5	0	0	
LO: Learning Objectives PQ: Program Qualifications																	
Contrib 1 ver ution Level:		very	low		2 low 3 M			Medi	dium 4 High			5 Very High					