	REACTIVE INTERMI	EDIAT	ES IN ORGANIC CHEMISTRY							
1	Course Title:	REACTI	VE INTERMEDIATES IN ORGANIC CHEMISTRY							
2	Course Code:	KIM5045								
3	Type of Course:	Optional								
4	Level of Course:	Second Cycle								
5	Year of Study:	1								
6	Semester:	1								
7	ECTS Credits Allocated:	6.00								
8	Theoretical (hour/week):	3.00								
9	Practice (hour/week):	0.00								
10	Laboratory (hour/week):	0								
11	Prerequisites:	None								
12	Language:	Turkish								
13	Mode of Delivery:	Face to t	face							
14	Course Coordinator:	GANİ KOZA								
15	Course Lecturers:	Prof. Dr. Necdet COŞKUN Prof. Dr. Mustafa TAVASLI								
16	Contact information of the Course Coordinator:	ganikoza@uludag.edu.tr +90 224 27 55 083 Uludağ Üniversitesi, Fen-Edebiyat Fakültesi, Kimya Bölümü, 16059 Görükle / BURSA, TÜRKİYE								
17	Website:									
18	Objective of the Course:	The aim of this course is to gain and teach students Carbocations, Carbens, Azetures, Nitrenes, Free Radicals, Singlet Oxygen and the formation and reaction of them.								
19	Contribution of the Course to Professional Development:	Recognize the intermediate products formed in organic reactions and make comments about the products formed.								
20	Learning Outcomes:									
		1	The student will learn the structure, stability, formation and reactions of carbocations							
		2	The student will know many named rearrangement reaction involving carbocation formation							
		3	The student will learn structure and reactivity of carbenes, synthesis and reaction of carbenes							
		4	The student will learn structure of azetures and nitrenes and formation and reactions of them							
		5	The student will learn structure of free radicals and reactions of them							
		6	The student will learn basic information about obtaining and reactions of singlet oxygen							
		7								
		8								
		9								
		10								
21	Course Content:									
		Co	ourse Content:							
Week	k Theoretical Practice									

1	Structure and stability of carbocations Formation and determination of carbocations Reactions of carbocations																
2	Rearrangement of carbocations: Wagner- Meerwein rearrangement. Classical carbocations																
3	Non-cl	assica	l carbo	cation	S											_	
4	Structu	re and	d reactiv	vity of	carben	IS											
5	Synthe	sis an	d react	ons o	f carbe	ns											
6	Rearra	ngem	ent of c	arben	S												
7	Midter	n Exa	m														
8	Structu nitrene	re and s	l forma	tion of	azetur	es an	d										
9	Curtius	and S	Schmid	h rear	ranger	nent											
10	Lessor	and I	lofman	n rear	rangen	nent											
11	Cycloa	dditior	n reaction	on of r	nitrenes	6											
12	Structu	re and	d stabili	ty of fr	ee rad	icals											
13	Formation and reactions of free radicals																
14	Produc	tion a	nd read	tions o	of singl	et oxy	gens										
22	Textbo	oks. R	eferen	ces an	d/or O	ther		1)	Organ	ic Che	mistrv.	Jonatha	an Clav	/den, Ni	ck Gree	ves.	
	Materia	als:						Stu	uart W	arren,	Peter V	/others,	Oxfor	d Chem	istry Pri	mer,	
Activites								1	Numb	er		Duration (hour)			Total Work Load (hour)		
									IGHT			3.00	3.00			42.00	
Practicals/Labs									0				0.00			0.00	
Self study and preperation									20,00 14				3.00			42.00	
Homeworks								6	6				8.00			48.00	
Projects									2000				0.00				
Field Studies									0				0.00			0.00	
Total Midterm exams									төр.оо				23.00			23.00	
Others									0			0.00			0.00		
Final Exams													25.00			25.00	
Total Work Load									<u> </u>						180.00		
Total work load/ 30 hr														6.00			
ECTS Credit of the Course									6.00								
24	ECTS	/ W0	ORK L	OAD	TAB	LE											
25			CON	TRIE	BUTIC	N O	F LE. (ARN QUA	ing (Lific	OUTC ATIO	COME: NS	S TO I	PROG	GRAM	ME		
	PG	1 PQ	2 PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16	
ÖK1	4	4	4	3	3	4	4	5	5	4	0	0	0	0	0	0	
ÖK2	5	4	3	3	4	3	5	4	3	3	0	0	0	0	0	0	
ÖK3	4	3	5	3	4	3	4	4	3	4	0	0	0	0	0	0	
ÖK4	4	5	4	5	3	4	4	5	4	5	0	0	0	0	0	0	

ÖK5	4	3	5	4	3	4	5	3	4	4	0	0	0	0	0	0
ÖK6	4	5	4	2	4	5	4	5	3	4	0	0	0	0	0	0
LO: Learning Objectives PQ: Program Qualifications																
Contrib 1 very low ution Level:				2 low			3 Medium			4 High			5 Very High			