TECHNICAL ENGLISH FOR CHEMISTS									
1	Course Title:	TECHNI	CAL ENGLISH FOR CHEMISTS						
2	Course Code:	KIM1004							
3	Type of Course:	Compuls	sory						
4	Level of Course:	First Cyc	cle						
5	Year of Study:	1							
6	Semester:	2							
7	ECTS Credits Allocated:	2.00							
8	Theoretical (hour/week):	2.00							
9	Practice (hour/week):	0.00							
10	Laboratory (hour/week):	0							
11	Prerequisites:	None							
12	Language:	Turkish							
13	Mode of Delivery:	Face to	face						
14	Course Coordinator:	Doç. Dr. MESUT GÖRÜR							
15	Course Lecturers:	Doç. Dr. Mesut GÖRÜR							
16	Contact information of the Course Coordinator:	mesutgorur@uludag.edu.tr							
17	Website:								
18	Objective of the Course:	The aim of the course is to enable students to have basic knowledge to master chemical terminology and express chemical processes concisely and fluently.							
19	Contribution of the Course to Professional Development:	Students will be able follow scientific literature and communicate written/verbal information in simple English sentences.							
20	Learning Outcomes:								
		1	Students are able to read, write and understand technical writings on chemistry in English.						
		2	Students are able to read and comprehend any technica report on chemistry written in English.						
		3	Students are able to use scientific vocabulary and terminology to give/write technical report on chemistry in English.						
		4	Students are able to read and write chemical procedu in English.						
		5							
		6							
		7							
		8							
		9							
		10							
21	Course Content:								
	Course Content:								
	Theoretical		Practice						
1	Introduction of the course (definition science, fields of science)								
2	Scientific writing 1 (chemistry, funda concepts in chemistry)	mental							

3	Scientific writing 2 (chemical units an	d						
	laboratory equipments)	u .						
4	Scientific writing 3 (Periodic table)							
5	Rephrasing 1 (states of matter)							
6	Rephrasing 2 (chemical reactions)							
7	Midterm Exam							
8	Organizing Information (Organic read	tions)						
9	Writing Styles in Chemistry (Thermochemistry)							
10	Discussion of Chemistry Journal Artic	cles 1						
11	Discussion of Chemistry Journal Artic	cles 2						
12	Discussion of Student Written Drafts							
13	Discussion of Student Written Drafts							
14	Discussion of Student Final Written Assignment							
22	Textbooks, References and/or Other Materials:		Course Book(s): Marting Bates, Tony Dudley-Evans, English for Science and Technology, Longman, 1983, N. A. Burnham, F. L. Hutson, "Scientific English as a Foreign Language", (2007) Basic English for Science, Oxford University Press 1978 http://www.upjs.sk/public/media/3499/English-for-Chemists.pdf Robinson, M.S.; Stoller, F.L.; Constanza-Robinson, M.S.; Jones, J.K. Write Like A Chemist Oxford University Press, New York. 2008. General Chemistry Principles and Modern Applications. Ninth Edition, Pearson International Edition. Petrucci, R.H.; Harwood, W.S.;					
23	Assesment							
TERM L	EARNING ACTIVITIES	NUMBE R	WEIGHT					
Midtern	n Exam	1	40.00					
Quiz		0	0.00					
Home v	vork-project	0	0.00					
Final E	xam	1	60.00					
Total		2	100.00					
Contribution of Term (Year) Learning Activities to Success Grade			40.00					
Contribution of Final Exam to Success Grade			60.00					
Total			100.00					
Measur Course		sed in the	Classic exams with multiple choice and/or open-ended questions					
24	ECTS / WORK LOAD TABLE							

Activites		Number	Duration (hour)	Total Work Load (hour)					
Theoretical		14	2.00	28.00					
Practicals/L	abs	0	0.00	0.00					
Self study a	and preperation	14	1.00	14.00					
Homeworks	8	0	0.00	0.00					
Projects		0	0.00	0.00					
Field Studie	es	0	0.00	0.00					
Midterm exa	ams	1	10.00	10.00					
Others		0	0.00	0.00					
Final Exam	s	1	10.00	10.00					
Total Work	Load			62.00					
Total work I	oad/ 30 hr			2.07					
ECTS Cred	it of the Course			2.00					
25 CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME									

25	CONTRIBUTION OF LEARNING OUTCOMES TO PROGRAMME QUALIFICATIONS															
	PQ1	PQ2	PQ3	PQ4	PQ5	PQ6	PQ7	PQ8	PQ9	PQ1 0	PQ11	PQ12	PQ1 3	PQ14	PQ15	PQ16
ÖK1	4	3	3	0	0	0	4	0	5	0	4	5	0	0	0	0
ÖK2	4	3	3	0	0	0	4	0	5	0	4	5	0	0	0	0
ÖK3	4	3	3	0	0	0	4	0	5	0	4	5	0	0	0	0
ÖK4	4	3	3	0	0	0	4	0	5	0	4	5	0	0	0	0
			LO: L	earr	ning (Objec	tive	s P	Q: P	rogra	ım Qu	alifica	tions	5	•	
Contrib ution Level:			2 low 3 N			Medi	lium 4 High			5 Very High						